

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
12 April 2001 (12.04.2001)

PCT

(10) International Publication Number
WO 01/25296 A1

- (51) International Patent Classification⁷: C08F 10/00 (74) Agent: FRANK B. DEHN & CO.: 179 Queen Victoria Street, London EC4V 4EL (GB).
- (21) International Application Number: PCT/GB00/03775
- (22) International Filing Date: 2 October 2000 (02.10.2000)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data:
99119575.1 2 October 1999 (02.10.1999) EP
- (71) Applicant (*for all designated States except US*): BOREALIS TECHNOLOGY OY [FI/FI]; P.O. Box 330, FIN-06101 Porvoo (FI).
- (72) Inventors; and
- (75) Inventors/Applicants (*for US only*): ERNST, Eberhard [DE/AT]; Borealis GmbH, Danubiastrasse 21-25, A-2323 Schwechat-Mannsworther (AT). REUSSNER, Jens [DE/AT]; Borealis GmbH, Danubiastrasse 21-25, A-2323 Schwechat-Mannsworther (AT). ROTERUD, Per, Torstein [NO/NO]; Borealis AS, N-3960 Stathelle (NO). SOLLI, Kjell-Arne [NO/NO]; Borealis AS, N-3960 Stathelle (NO).
- (81) Designated States (*national*): AE, AG, AL, AM, AT, AT (utility model), AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, CZ (utility model), DE, DE (utility model), DK, DK (utility model), DM, DZ, EE, EE (utility model), ES, FI, FI (utility model), GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KR (utility model), KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SK (utility model), SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.
- (84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

Published:

— With international search report.

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.



WO 01/25296 A1

(54) Title: MODIFIED SUPPORTED CATALYSTS FOR THE POLYMERIZATION OF OLEFINS

(57) Abstract: A modified supported olefin polymerization catalyst, characterised in that the catalyst is modified by pre-polymerization with an olefin or an olefin mixture different from the olefin or olefin mixture of the subsequent olefin polymerization.

BEST AVAILABLE COPY

Modified Supported Catalysts
for the Polymerization of Olefins

The invention relates to modified supported catalysts for the polymerization of olefins, for example based on single-site and multi-site metallocene catalysts, supported on inorganic or organic supports, or based on the Ziegler-Natta type catalysts.

Metallocenes of the metals of e.g. transition group IV of the Periodic Table of the elements are highly active catalysts for the polymerization of olefins. The resulting polyolefins have new advantageous combinations and supplement the product range of the polyolefins prepared hitherto using known conventional Ziegler-Natta catalysts.

It is known that catalysts based on unbridged, substituted and unsubstituted biscyclopentadienyl metallocenes in combination with aluminoxanes as cocatalyst can be used for the preparation of polyethylene and ethylene-olefin copolymers (EP 0 128 046).

It is also known that stereoregular polyolefins can be prepared using bridged, chiral metallocenes. For bridging the ligand systems, use is mostly made of dimethylsilanediyl groups (EP 0 316 155), methylphenylsilanediyl groups (EP 0 320 762), ethylene groups (Brintzinger, J. Organomet. Chem. 288 (1985), 63-67) isopropylidene bridges (EP 0 459 264) and silyl-substituted diyl bridges (WO 97 02 276).

Depending on the ligand type and the substituents, isotactic, syndiotactic, hemiisotactic, stereoblock-type and atactic homopolymers and copolymers having aliphatic

- 2 -

or cyclic structures can be prepared.

As ligands, preference is given to using substituted and unsubstituted cyclopentadienyl units (EP 316 155), substituted and unsubstituted indenyl units (EP 0 302 424; EP 0 485 823) and also substituted and unsubstituted cyclopentadienyl units in combination with unsubstituted fluorenyl groups (EP 0 412 416).

It is further known that bridged metallocenes having a cyclopentadienyl system and a heteroatom ligand (constrained geometry catalyst) can also be used for the polymerization of olefins (US 5 096 867).

A disadvantage of such homogeneous catalysts in the polymerization of olefins are the resulting powders, having only a low bulk density. The particle morphology of such products can in principle be somewhat improved by a specific pretreatment of the metallocene with the cocatalyst (EP 0 302 424). However, such a process has the disadvantage of, in particular, heavy deposit formation in industrial reactors (EP 0 563 917).

Although the use of methylaluminoxane, which is insoluble in aliphatic solvents, as support material gives a certain improvement in the activity and likewise leads to pulverulent products [Polymer 32(1991), 2671-2673].

Supporting the metallocene on oxidic materials such as silicon oxide or aluminium oxide with pretreatment of the starting material, which may be partially dehydrated, with the cocatalyst is a known method (WO 91 09 882) used in homopolymerization and copolymerization of ethylene. However, in this method, the particle size of the polymer particle is determined essentially by the particle size of the support material.

- 3 -

Further processes describe the modification of the oxidic support using methylaluminoxanes and the subsequent application of the metallocene (EP 0 206 794). However, this method restricts the ability to control the particle size by means of the properties of the support materials.

EP 0 685 494 describes a further supported catalyst which is prepared by the application of methylaluminoxane to a hydrophilic oxide, subsequent crosslinking of the methylaluminoxane using a polyfunctional organic crosslinker and subsequent application of an activated methylaluminoxane/metallocene complex. The disadvantage of this supported catalyst is that at the relatively high polymerization conversions achieved in industrial plants, the strength of the supported catalysts is not sufficient to ensure a compact, granular morphology of the polymer product.

A support material by drying hydrophilic inorganic oxides and reacting with aluminosilicates and polyfunctional crosslinkers gives, after contacting with catalysts, a supported catalyst system for a stable high bulk density in olefin polymerization even at high conversion rates (EP 0 787 746).

Methods for modifying homogeneous catalysts and supported catalyst systems by pre-polymerization are also known.

In EP 0 354 893 a homogeneous catalyst is prepared by pre-polymerizing a precipitated complex of an aluminium alkyl and a metallocene catalyst with an olefin monomer at a temperature below the polymerization temperature of the monomer, followed by polymerizing the olefin monomer under polymerization conditions. EP 0 426 638 describes the pre-polymerization of propylene in combination with

- 4 -

homogeneous metallocene catalysts for reproducible and controllable polymerization. The disadvantage of homogeneous polymerization, resulting in powders having only a low bulk density, is not resolved in these processes.

For supported catalyst systems, pre-polymerization of olefinic monomers is known from EP 0 705 281 using bis-indenyl metallocene reaction product catalysts whereby the indenyl rings are 2-substituted, further from WO 97 02 297 wherein the pores of the catalyst system contain a volume of liquid that is equal to or less than the total pore volume of the supported catalyst system. In WO 96 28 479, a supported metallocene catalyst system having an activity greater than 100,000 g/g/h and at least an α -olefin monomer are combined under pre-polymerization conditions, hydrogen is added, and after recovering the prepolymerized supported catalyst system, ethylene or propylene is fed for polymerization.

However, these pre-polymerizations generally use the same olefin or olefin mixture as the olefin or olefin mixture of the subsequent olefin polymerization, or in the case of WO 99 24 478 the pre-polymerized catalyst is used directly in the subsequent polymerization.

One of the main disadvantages of these known pre-polymerization processes is the impossibility to use the known metallocene supported catalysts necessary for the production of high molecular weight polyolefins without reactor fouling, formation of polymer fines resulting from soluble catalyst components and disrupted catalyst components in the initial phase of the olefin polymerization under technical liquid olefin polymerization conditions.

It is therefore an object of the invention to find a

- 5 -

supported catalyst for liquid-phase polymerization of olefins for the production of high molecular weight polyolefins without reactor fouling, formation of polymer fines resulting from soluble catalyst components and disrupted catalyst components in the initial phase of the olefin polymerization.

It has now surprisingly been found, that these requirements are achieved by a modified supported olefin polymerization catalyst, characterised in that the catalyst is modified by pre-polymerization with an olefin or an olefin mixture different from the olefin or olefin mixture of the subsequent olefin polymerization.

The supported catalyst may be a metallocene catalyst, for example with one, two or three η^5 ligands bonding to a transition metal or lanthanide, preferably one or two η^5 ligands such as cyclopentadienyl, indenyl or fluorenyl. The η^5 ligands may be linked to each other by bridging groups, for example to yield a bridged bis- η^5 ligand. Where only one η^5 ligand is bonded to the metal, a pendant group on the ligand can also form a σ bond to the metal, a so-called "scorpionate" catalyst.

Ziegler-Natta catalysts are well known, and the invention may also be useful with other catalysts such as Philips (CrO_x) catalysts.

Metallocene catalysts may be classified as single-site or multi-site. Both may be coactivated by a cocatalyst, for example methyl aluminoxane, and may be supported on an inorganic or organic carrier. Ziegler-Natta catalysts, which are well known as multi-site catalysts, are supported by their nature.

The olefin used for the polymerization, different from the olefin of the subsequent polymerization, may be one

- 6 -

or more of propylene, 1-butene, 3-methyl-1-butene, 3,3-dimethyl-1-butene, 3-methyl-1-pentene, 3-methyl-1-hexene, 3,4-dimethyl-1-pentene, 3,4-dimethyl-1-hexene, 1-pentene, 4-methyl-1-pentene, 4-methyl-1-hexene, 4,5-dimethyl-1-hexene, 4,4-dimethyl-1-pentene, 4,4-dimethyl-1-hexene, 6,6-dimethyl-1-heptene, vinyl cyclopropane, vinyl cyclobutane, vinyl cyclopentane, vinyl cyclohexane, vinyl cycloheptane, vinyl cyclooctane, allyl cyclopropane, allyl cyclobutane, allyl cyclopentane, allyl cyclohexane, allyl cycloheptane or allyl cyclooctane.

Preferred are one or more of propylene, 1-butene, 3-methyl-1-butene, 3,3-dimethyl-1-butene, 4-methyl-1-pentene, 4,4-dimethyl-1-pentene, 4,5-dimethyl-1-hexene, vinyl cyclopentane, vinyl cyclohexane, vinyl cycloheptane, allyl cyclopentane, allyl cyclohexane or allyl cycloheptane, whilst particularly preferred olefins are 3-methyl-1-butene, 4-methyl-1-pentene, vinyl cyclopentane and vinyl cyclohexane.

Typically the pre-polymerization will result in 0.001 - 5.0, e.g. 0.001 - 3.0 or 0.01 - 1.0 parts by weight relative to the catalyst and support of the polyolefin or olefin copolymer different from the polyolefin or olefin copolymer of the subsequent olefin polymerization.

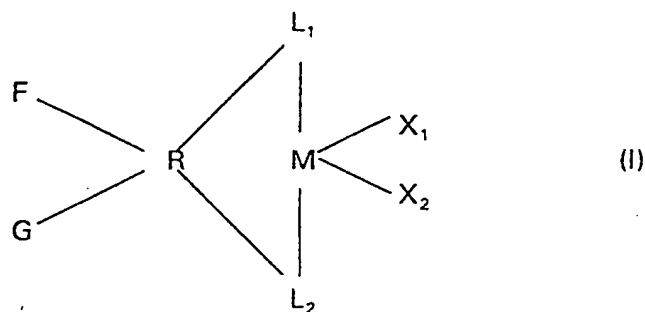
The melting point or softening point of the polyolefin or olefin copolymer resulting from pre-polymerization is suitably at least 20°C, preferably 30°C higher than the melting point or softening point of the polyolefin or olefin copolymer of the subsequent olefin polymerization.

One particular supported metallocene catalyst for olefin polymerization which may be modified according to the

- 7 -

invention, especially for liquid-phase polymerization of olefins, comprises

- A) 90.0 - 99.9 parts by weight of a catalyst support based on a hydrophilic inorganic oxide of an element of main groups II to IV or transition group IV of the Periodic Table or a mixture or mixed oxide thereof, which catalyst support is obtainable by simultaneous reaction with aluminoxanes and with polyfunctional organic crosslinkers,
- B) 10-0.1 parts by weight of a metallocene compound of the formula I



where

M is a metal selected from the group consisting of Ti, Zr, Hf, V, Nb and Ta or an element selected from the group consisting of the lanthanides,
 X_1 and X_2 are identical or different and are each a C_1 - C_{10} -alkyl group, a C_1 - C_{10} -alkoxy group, a C_6 - C_{10} -aryl group, a C_6 - C_{10} -aryloxy group, a C_2 - C_{10} -alkenyl group, a C_7 - C_{20} -alkylaryl group, a C_7 - C_{20} -arylalkyl group, a C_8 - C_{20} -arylalkenyl group, hydrogen or a halogen atom,

L_1 and L_2 are identical or different and are each an unsubstituted, monosubstituted or polysubstituted monocyclic or polycyclic hydrocarbon radical containing at least one cyclopentadienyl unit which can form a sandwich structure with M, R is carbon, silicon, germanium or tin, F and G are identical or different and

- 8 -

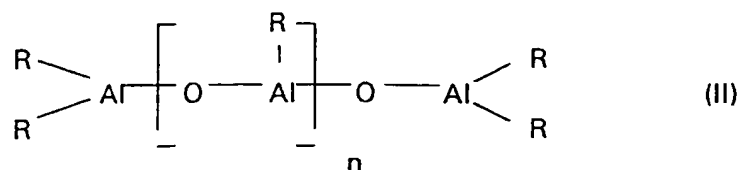
are each a trimethylsilyl radical of the formula -
 $\text{Si}(\text{CH}_3)_3$, where G may also be a C_1 - C_{10} -alkyl radical,
preferably a C_1 - C_4 -alkyl radical, or a C_6 - C_{10} -aryl
radical,

whereby according to the invention said modified
supported metallocene catalyst comprising the catalyst
support A and the metallocene B has an activity of
maximum 100 kg, preferably maximum 50 kg, particularly
preferred maximum 20 kg polyolefin/g supported
catalyst/h. It is also possible that the supported
metallocene catalyst has an activity of maximum 5 kg or
maximum 10 kg polyolefin/g supported catalyst/h.

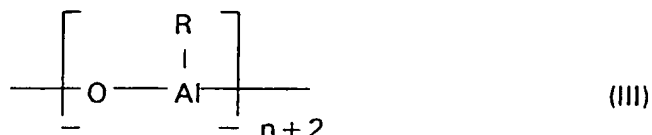
The hydrophilic oxides used in the catalyst support
usually contain hydroxyl groups and/or physically
absorbed water. They are preferably porous and finely
divided and usually have a mean particle size of from 10
to 300 microns.

Preferably the hydrophilic inorganic oxide is an
aluminum oxide (alumina), silicon oxide (silica),
magnesium oxide, titanium oxide or zirconium oxide or a
mixture or mixed oxide thereof. Particular preference is
given to using silicon dioxides of the Grace Davison
type. However, other suitable starting materials are
finely divided oxides, for example those described in EP
0 585 544, which are prepared by high temperature
hydrolysis from gaseous metal chlorides or silicon
compounds. Magnesium chloride is another suitable
support material.

According to the present invention, the aluminoxane used
in the catalyst support is a linear aluminoxane of the
formula II



or an aluminoxane of the cyclic type III



where, in the formula II and III, the radicals R can be identical or different and are each a C₁-C₆-alkyl group and n is an integer in the range 1 - 50. Preferably, the radicals R are identical and are methyl, isobutyl, phenyl or benzyl. The aluminosiloxane can be prepared in various ways by known methods. One possibility is, for example, the reaction of aluminium alkyls with aluminium sulphate containing one water of crystallisation (EP 0 302 424).

In the catalyst support, the molar ratio of aluminium (as aluminoxane) to surface hydroxyl groups of the hydrophilic inorganic oxide is between 1 and 50, preferably between 1 and 30.

According to the invention, suitable polyfunctional organic crosslinkers in the catalyst support are all organic compounds having more than one functional group which can react with metal-carbon bond. Preference is given to using a bifunctional crosslinker. Such bifunctional organic compounds can be, for example, aliphatic or aromatic diols, aldehydes, dicarboxylic

- 10 -

acids, primary or secondary diamines, diepoxy compounds. Most preference is given to using aliphatic and aromatic diols, secondary amines or diepoxy compounds or mixtures thereof. Particular preference is given to using ethylene glycol, butanediol, bisphenol A and 1,4-butanediol diglycidyl ether. Tri- or higher functional crosslinkers which can be used are, for example, triethanolamine, glycerol, phloroglucinol or tetraethylenepentamine.

The molar ratio between the aluminium as aluminoxane and the polyfunctional organic compound in the catalyst support can vary within a wide range and is between 1 and 100, preferably between 1 and 40. Higher molar ratios of aluminium to polyfunctional organic compound are used particularly when use is made of tri- or higher-functional crosslinkers which can form a correspondingly higher number of crosslinks.

According to the invention, preferred ligands L_1 and/or L_2 in the metallocene compound are substituted or unsubstituted cyclopentadienyl, indenyl or fluorenyl radicals. Particular preference is given to cyclopentadienyl, tetramethylcyclopentadienyl, indenyl, 2-methylindenyl, 2-methyl-4-phenylindenyl, 2-methyl-4,5-benzindenyl and fluorenyl units and also ferrocene- and ruthenocene-substituted units as are described, for example, in EP 0 673 946.

According to the invention, the following metallocenes are particularly preferred:

bis(trimethylsilyl) silanediylcyclopentadienyl-zirconium dichloride,

bis(trimethylsilyl) silanediylindenylzirconium dichloride,

bis(trimethylsilyl) silanediylbis (2-methylindenyl) zirconium dichloride,

- 11 -

bis(trimethylsilyl) silanediylbis (2-methyl-4, 5-benzoindenyl) zirconium dichloride,
bis(trimethylsilyl) silanediylbis (2-methyl-4-phenylindenyl) zirconium dichloride,
bis(trimethylsilyl) silanediylbis (2-methyl-4-naphthylindenyl) zirconium dichloride,
bis(trimethylsilyl) silanediyl difluorenylzirconium dichloride,
bis(trimethylsilyl) silanediyl (fluorenyl) (cyclopentadienyl) zirconium dichloride,
bis(trimethylsilyl) silanediyl (fluorenyl) (indenyl) zirconium dichloride,
bis(trimethylsilyl) silanediyl (tetramethylcyclopentadienyl) (indenyl) zirconium dichloride,
methyl (trimethylsilyl) silanediyl dicyclopentadienylzirconium dichloride,
methyl (trimethylsilyl) silanediyl diindenylzirconium dichloride,
methyl (trimethylsilyl) silanediylbis (2-methylindenyl) zirconium dichloride,
methyl (trimethylsilyl) silanediylbis (2-methyl-4,5-benzoindenyl) zirconium dichloride,
methyl (trimethylsilyl) silanediylbis (2-methyl-4-phenylindenyl) zirconium dichloride,
methyl (trimethylsilyl) silanediylbis (2-methyl-4-naphthylindenyl) zirconium dichloride,
methyl (trimethylsilyl) silanediyl difluorenylzirconium dichloride,
methyl (trimethylsilyl) silanediyl (fluorenyl) (cyclopentadienyl) zirconium dichloride,
methyl (trimethylsilyl) silanediyl (fluorenyl) (indenyl) zirconium dichloride and
methyl (trimethylsilyl) silanediyl (tetramethylcyclopentadienyl) (indenyl) zirconium dichloride.

According to the invention, in the metallocene compound in addition also amido, phosphido and arsenido radicals

- 12 -

can be used as ligands L_2 , where the substituents of these ligands are as defined for X_1 and X_2 , or substituted or fused ferrocenyl- or ruthenocenyl-radicals.

The invention further provides a process for preparing the modified supported catalyst as hereinbefore described, comprising the step of modifying a supported olefin polymerization catalyst by pre-polymerization with an olefin or an olefin mixture different from the olefin or olefin mixture of the subsequent olefin polymerization, whereby the melting point or softening point of the polyolefin or olefin copolymer resulting from pre-polymerization is at least 20°C, preferably 30°C higher than the melting point or softening point of the polyolefin or olefin copolymer, of the subsequent olefin polymerization.

The modified catalyst may be isolated from the modification medium prior to use in the subsequent polymerization reaction.

In particular, a process for preparing a modified supported metallocene catalyst for polymerization of olefins, especially in liquid phase, comprises the steps:

α) preparing a catalyst support A) by

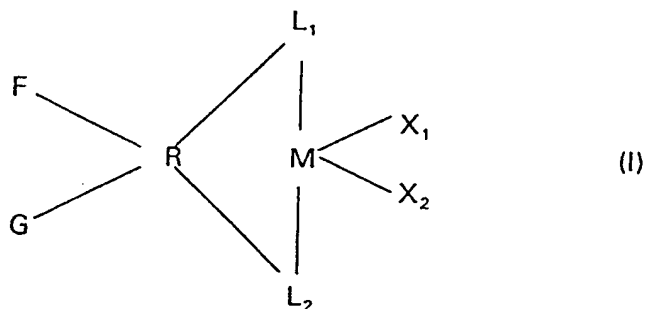
α1) drying a hydrophilic inorganic oxide of an element of main groups II to IV or transition group IV of the Periodic Table or a mixture or mixed oxide thereof at from 110 to 900°C, e.g. 110 to 800°C, subsequently

α2) if desired, reacting the free hydroxyl groups of the oxide completely or partially with aluminoxanes or aluminium alkyls and subsequently

- 13 -

α3) reacting the oxide simultaneously with aluminoxanes and polyfunctional organic crosslinkers,

β) suspending the catalyst support A) in an inert hydrocarbon and bringing it in contact with a solution of a metallocene compound B) of the formula I in an inert hydrocarbon, wherein 90.0 - 99.9 parts by weight of a catalyst support A) are mixed with 10 - 0.1 parts by weight of the metallocene compound B)



where

M is a metal selected from the group consisting of Ti, Zr, Hf, V, Nb and Ta or an element selected from the group consisting of the lanthanides,
 X_1 and X_2 are identical or different and are each a C_1 - C_{10} -alkyl group, a C_1 - C_{10} -alkoxy group, a C_6 - C_{10} -aryl group, a C_6 - C_{10} -aryloxy group, a C_2 - C_{10} -alkenyl group, a C_7 - C_{20} -alkylaryl group, a C_7 - C_{20} -arylalkyl group, a C_8 - C_{20} -arylalkenyl group, hydrogen or a halogen atom,

L_1 and L_2 are identical or different and are each an unsubstituted, monosubstituted or polysubstituted monocyclic or polycyclic hydrocarbon radical containing at least one cyclopentadienyl unit which can form a sandwich structure with M, R is carbon, silicon, germanium or tin,

F and G are identical or different and are each a trimethylsilyl radical of the formula $-Si(CH_3)_3$, where G may also be a C_1 - C_{10} -alkyl radical, preferably a alkyl

- 14 -

radical, or a C₆-C₁₀-aryl radical,

γ) modifying said supported metallocene catalyst as defined above.

In the first stage α1), the oxide is preferably dehydrated in a stream of nitrogen or under reduced pressure at temperatures of from 110 to 800°C or 900°C over a period from 1 to 24 hours. The concentration of free hydroxyl groups, established as a function of the drying temperature selected, is then measured. The free hydroxyl groups can be reacted completely or partially with aluminoxanes or aluminium alkyls in stage α2).

In stage α3), the dried oxide is reacted simultaneously with aluminoxanes and at least one polyfunctional organic crosslinker, which is suspended, for example, in a suitable hydrocarbon solvent such as toluene in such a way that it is covered with the solvent. The solvents for the aluminoxane and for the crosslinker have to be miscible and the same solvents are preferable used. Particular preference is given to using toluene.

To prepare the solution needed in stage α3), the solvent used for the crosslinker can be the same as for the aluminoxane solution. Owing to the temperature dependence of the solubility of these crosslinkers in the solvent used, the desired concentration can be set in an targeted manner by the choice of the temperature of the solution. Particularly advantageous is the selection of a solvent whose boiling point is below the decomposition temperature of the solid prepared in stage α3). Preference is given to using aromatic solvents such as xylene, benzene or toluene.

When using the polyfunctional crosslinkers in preparing the catalyst support in stage α3), it is also possible,

- 15 -

in a further reaction stage, to deactivate unreacted reactive groups using, for example, alkylaluminium compounds, preferably using trimethylaluminium.

The molar ratio between the aluminium used in stage $\alpha 3$) as aluminosilane and the crosslinker can vary within a wide range and is between 1 and 100, preferably between 1 and 40, particularly preferably between 10 and 25. It is dependent, in particular, on the type and pretreatment of the metal oxides, the type of aluminosilanes used, on the respective molar ratio of aluminium (as aluminosilane) to the surface hydroxyl groups on the metal oxide and on the type of crosslinker. Higher molar ratios of aluminium to the crosslinker are used particularly when use is made of tri- or higher functional crosslinkers which can form a correspondingly higher number of crosslinks.

The suspended dried oxide from stage $\alpha 1$) is preferably treated with a solution of aluminosilane and a solution of one or more polyfunctional organic crosslinkers in the same solvent. If desired, it is also possible in stage $\alpha 2$) to react the free hydroxyl groups of the oxide with an up to equimolar amount of an aluminosilane or an aluminium alkyl solution, for example trimethylaluminium prior to the crosslinking reaction. Preference is given to using methylaluminosilane for this purpose. It has been found to be particularly advantageous if all hydroxyl groups have been reacted. However, even a partial reaction of these groups gives a positive effect.

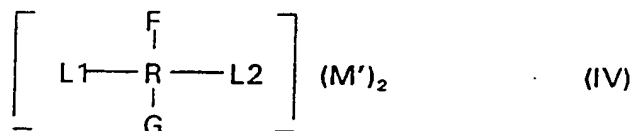
According to the invention, in stage $\alpha 3$) for preparing the catalyst support, the metering in of the solutions is carried out simultaneously and continuously and the crosslinker solution may be heated or cooled if desired. The temperature to which the solution is heated or

- 16 -

cooled depends on the solubility of the crosslinker in the solvent selected and on the desired crosslinking density on the support surface. The rate at which the two streams are metered in can be set by means of metering pumps and is in a range between 0.1 and 1000 ml per minute, preferably between 0.5 and 250 ml per minute, particularly preferably between 1 and 50 ml per minute. The reaction is preferably carried out in such a way that all the aluminosilane has been reacted after the simultaneous metering in of the two solutions. Under some circumstances, fluctuations in the reaction conditions on the industrial scale can lead to unreacted aluminosilane remaining in the solution. The usable catalyst supports as described in EP 0 685 494 display a soluble aluminium proportion in the solvent used of preferably less than 1.4 mol% based on the aluminosilane used. In this case, it is possible to carry out one or more washing steps in order to reduce the concentration below the desired limit. It is also possible to add a further amount of aluminosilane and to store the resulting suspension to improve the catalyst - performance (DE-OS 19821370).

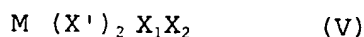
After addition of the reactants in stage $\alpha 3$) for preparing the catalyst support is complete, the reaction mixture is stirred further for about 60 minutes and the solvent is then removed. The residue can be dried under reduced pressure, but it is preferably used further in the moist state.

The metallocene compounds, used for the supported metallocene catalyst B are prepared by reacting metallocenes of the formula IV



- 17 -

where L_1 , L_2 , F, G and R are as defined for the formula I and M' is an alkali metal, preferably lithium, with a compound of the formula V



Where M, X_1 and X_2 are as defined for the formula I and X' is a halogen atom, preferably chlorine.

Preparing the metallocene compounds B), the reaction of the dimetallated compound of the formula IV with the metal halide of the formula V can be carried out, for example, as described in EP 0 659 756. However, the reaction of the dimetallated compound of the formula IV with the metal halide of the formula V is advantageously carried out in solvent mixtures of aromatic and/or aliphatic hydrocarbons, which may also be halogenated, with dialkyl ethers, preferably alkane/ether mixtures such as, for example, hexane/ether mixtures. Aliphatic hydrocarbons can be, for example, all C_5 - C_{12} -alkanes. Preference is given to n-pentane, n-hexane, n-heptane or cyclohexane. Among the dialkylethers, preference is given to all di- C_2 - C_4 -alkyl ethers, for example diethyl ether, di-n-propyl ether, diisopropyl ether, di-n-butyl ether or tert-butyl-methyl ether. Examples of suitable halogenated hydrocarbons are all C_1 - C_4 -chloroalkanes. Particular preference is given to dichloromethane.

The preparation of the supported metallocene catalyst according to β is carried out by suspending the catalyst support A) in an inert hydrocarbon, preferably toluene, and bringing it into contact with the metallocene compound B). In this procedure, the metallocene compound is dissolved, for example, in an inert hydrocarbon. Inert solvents which can be used are, for example, aliphatic or aromatic hydrocarbons, preferably toluene. The metallocene compounds B) are preferably

- 18 -

used in an amount of from 0.3 wt % to 5 wt % based on the total mass of the supported catalyst. The mixing time is from 5 minutes to 24 hours, preferably from 1 to 12 hours. The mixing is carried out at a temperature of from -10 to +80°C, in particular from 20 to 70°C. The application of the metallocene compound B) is preferably carried out subsequent to the synthesis of the support A) in order to save a drying step. After the reaction is complete, the solvent is decanted and taken off under reduced pressure until a free-flowing solid remains.

According to the invention the pre-polymerization of the supported catalyst is carried out in absence of a solvent or in an inert hydrocarbon solvent. The pre-polymerization temperature is from particularly -20 to 95°C, preferably from 0 to 60°C. The treatment of the supported catalyst may be carried out under reduced, atmospheric or elevated pressure batchwise or continuously.

In a preferred embodiment the monomer feed during pre-polymerization reaction is at a rate from 0.05 to 20 g monomer/g supported catalyst/h, more preferably from 0.5 to 2 g monomer/g supported catalyst/h.

According to the invention, the pre-polymerization of these monomers in presence of the supported catalyst is carried out preferably in gas phase, although pre-polymerization by pore filling of a dry porous catalyst with liquid monomer, by slurry phase polymerization using oil or grease as a slurry medium or by slurry phase polymerization using an evaporatable hydrocarbon as slurry medium are also possible.

In a preferred embodiment especially for liquid-phase polymerization of propylene, the modification of the

- 19 -

supported metallocene catalyst is carried out by pre-polymerization of 4-methyl-1-pentene.

The modified supported catalyst of the invention is polymerization-active without further activating additives. However, it has been found to be particularly advantageous to use aluminium alkyls, preferably trimethylaluminium, triethylaluminium or triisobutylaluminium, as scavenger and as additional activator. The amount is, based on the aluminium, 50 - 5000 mol, preferably 100 - 500 mol, per mol transition metal of the metallocene compound in the modified supported catalyst.

According to the invention the modified supported catalysts may be used for the polymerization of olefins or olefin mixtures, in particular C_2 - C_{16} - α -olefins, preferably ethylene, propylene, 1-butene, 1-hexene, 1-octene, 4-methyl-1-pentene, cyclopentene, norbornene and/or norbornadiene. The modified supported catalysts make possible the preparation of homopolymers, copolymers and heterophasic copolymers. Most preferred is the use of the modified supported catalysts for the polymerization of propylene.

The invention further provides a process for preparing polyolefins by polymerization or copolymerization of olefins, wherein the polymerization catalyst used is a modified supported catalyst according to the invention.

In liquid phase polymerization, it is possible to use inert solvents, for example, aliphatic or cycloaliphatic hydrocarbons such as pentane, hexane or cyclohexane; toluene can also be used. Preference is given to carrying out the polymerization in the liquid monomer.

If required, hydrogen is added as molecular weight

- 20 -

regulator. The total pressure in the polymerization is usually 0.5 - 150 bar. Polymerization is preferably carried out in the pressure range of 1 - 40 bar.

In the copolymerization of ethylene with propylene, the polymerization is preferably carried out in liquid propylene or in hexane as suspension medium. In the polymerization in liquid propylene, the ethylene is preferably fed in an amount such that a partial pressure ratio $P_{\text{propylene}}/P_{\text{ethylene}} > 0.5$, in particular > 1.0 , is established over the liquid phase (P_{ethylene} = partial pressure of the ethylene in the gas phase over the suspension; $P_{\text{propylene}}$ partial pressure of the propylene in the gas phase over the suspension). In the copolymerization in hexane as suspension medium, an ethylene/propylene gas mixture having a propylene content of from 1 to 50 mol%, preferably from 5 to 30 mol%, is fed in. The total pressure is kept constant during the polymerization by metering in a further amount. The total pressure is from 0.5 to 40 bar, preferably from 1 to 20 bar. The polymerization time is from 10 minutes to 6 hours, preferably from 30 minutes to 2 hours.

The modified supported catalysts make possible the preparation of high molecular weight polyolefins and olefin copolymers without reactor fouling, without formation of polymer fines resulting from soluble catalyst components and disrupted catalyst components in the initial phase of the olefin polymerization, especially under technical liquid polymerization conditions. Particle size and particle size distribution may be controlled in all stages of polymerization process. The polyolefins and olefin copolymers are obtained in granular form even at high polymerization conversions, they have a high bulk density and a low fines content.

The following examples illustrate the invention.

Abbreviations used are:

$^{13}\text{C-NMR}$	^{13}C nuclear magnetic resonance spectroscopy
d_{50}	Mean particle diameter determined by sieve analysis
$^1\text{H-NMR}$	^1H nuclear magnetic resonance spectroscopy
M_w	Weight average molar mass in g/mol determined by GPC
M_n	Number average molar mass in g/mol determined by GPC
M_w/M_n	Polydispersity
MAO	Methylaluminoxane
MC	Metallocene compound MS mass spectrometry
T_m	Melting point determined by DSC
cat_{pure}	supported catalyst without pre-polymer
$\text{cat}_{\text{pre-pol}}$	p re-polymerized catalyst
m_{cat}	weight of the supported catalyst
$m_{\text{pre-pol}}$	weight of the pre-polymer
TEAL	triethyl aluminium
MFR	melt flow rate of PP in g/10 min measured at 230°C using a weight of 2.16 kg
MFR_2	melt flow rate of PE in g/10 min measured at 190°C using a weight of 2.16 kg
MFR_{21}	melt flow rate of PE in g/10 min measured at 190°C using a weight of 21.6 kg
XS	amount of polymer still soluble in xylene at 25°C after precipitation from high temperature solution

- 22 -

Example 1

Preparation of the catalyst support

The silica (pore volume 1.58 cm³/g; surface area 300 m²/g) was dried at 200°C using the procedure described in EP 787 746 Example 1. The OH content was 2.05 mmol/g SiO₂.

109.1 g of the dried oxide were suspended in 1.41 of dry toluene in a three-neck flask equipped with a stirrer and two dropping funnels. In the first dropping funnel, 252 g (1298.3 mmol of Al) of a 30% strength solution of methylaluminoxane (MAO) in toluene were mixed with 740 ml of toluene and the second dropping funnel was charged with a solution of 12.25 g bisphenol A (53.72 mmol) in 613 ml oxygen-free toluene.

After 127.5 ml of the MAO solution had been initially introduced into the suspension at 55°C within 15 minutes under moderate stirring, the two solutions were simultaneously added dropwise within approximately 40 minutes. The stirrer speed during this procedure was 200 rpm. The rate of dropwise addition was low and was selected such that both solutions were consumed at a suitable decrease in volume. The suspension was subsequently stirred for further three hours at 55°C and was then allowed to stand. A fine white solid settled out. After pressurizing with 0.3 bar nitrogen, the cold suspension was stored for at least three weeks.

Subsequently the supernatant toluene was removed, the residue was again taken up two times with 800 ml toluene and washed for 15 minutes at 70°C. After separation of the white solid the supernatant washing solutions were removed. The resulting catalyst support suspension was directly used for the catalyst preparation.

- 23 -

Example 2

Preparation of the metallocene compound

121 ml of a 1.5 molar methyllithium solution (180.3 mmol) are added under argon at -80°C to a solution of 47.5 g (90.2 mmol) of 1,1,1,2-tetramethylbis [2-methyl-4-phenylindenyl] disilane (M. Kumada, J. Organomet. Chem. 43 (1972) 2931 in 400 ml of absolute diethyl ether. The reaction mixture is brought to room temperature over a period of 1 hour and stirred further for 20 hours at room temperature. After taking off the solvent in a high vacuum, a red-brown oily residue is obtained. This is admixed with 200 ml of absolute n-hexane, the resulting suspension is filtered under argon and the filtration residue is washed three times with absolute n-hexane. After drying for 30 minutes in a high vacuum, 46.6 g of the corresponding dilithium salt (96.0 % yield) is obtained as an orange powder which is used without further purification for the synthesis of the zirconocene dichloride.

A suspension of 21.0 g of zirconium tetrachloride in 300 ml of absolute n-hexane is added at -80°C to a solution of 46.6 g of the dilithium salt in 300 ml of absolute diethyl ether. After warming to room temperature over a period of two hours, the mixture is stirred for 20 hours at room temperature. After taking off the solvent in a high vacuum, the solid residue is washed twice with 100 ml absolute diethyl ether. The ether fractions are collected, the solvent is evaporated to a residual volume of 100 ml and the solution is cooled to -30°C for a few days, yielding an amorphous product which is isolated by filtration. After 3 hours drying in vacuum, a 10.0 g amount of pseudoracemethyl(trimethylsilyl)-silanediyl(2-methyl-4-phenylindenyl) zirconium dichloride (16.0 % yield) is obtained. After further recrystallization from diethylether at 30°C a purity of

- 24 -

99 % (11.0 % yield) is reached.

$C_{36}H_{36}Si_2ZrCl_2$ (686.99 g/mol)

MS (EI, 30 eV) [m/z (%]): 687 (14)

1H -NMR (CD_2Cl_2) [ppm]: 0.62 (s, 9H, $(CH_3)_3Si-SiCH_3$); 1.51 (s, 3H, $(CH_3)_3Si-SiCH_3$); 2.26 and 2.32 (s, 3H, CH_3 -Ind); 6.95 to 7.62 (m, 18 H, Ph/Ind-H)

Example 3

Preparation of the supported metallocene catalyst

To 19.96 g of the catalyst support, prepared according to example 1, suspended in toluene (the solid is covered), a solution comprising 0.177 g of methyl (trimethylsilyl) silanediylbis (2-methyl-4-phenylindenyl) zirconium dichloride, prepared according to Example 2 and 60 ml of toluene was added. The suspension became orange coloured and this colour became increasingly intense during the course of the stirring phase. After 16 hours, the stirrer was switched off. The catalyst was washed 2 times and the supernatant clear solution was decanted. Removal of the solvent at 50°C under reduced pressure gave a red/orange finely divided solid. The metallocene content was 0.9 wt %.

Example 4

Propylene polymerization using the supported metallocene catalyst

A 20 l stirred reactor was, after being made inert, charged at room temperature with 6.5 kg of liquid propylene and the mixture was stirred for 15 minutes at 200 rpm.

438 mg of the supported metallocene catalyst prepared in Example 3 were placed in the catalyst feeder.

- 25 -

Subsequently the powder was rinsed into the reactor using a further 500 g of propylene, the stirrer speed was increased to 400 rpm, the mixture was heated over a period of 12 minutes to the polymerization temperature of 70°C and the temperature was kept constant. The reaction was stopped after two hours by flashing off the propylene. This gave 109.5 g of polypropylene with a fines content (<100 μm) determined by sieve analysis of 3.8 % by weight. The activity was $0.25 \text{ kg/g cat}_{\text{pure}} \times 2 \text{ h}$. The polymer particles were granular ($M_w = 604 \text{ kg/mol}$; polydispersity 2.9; $T_m = 153^\circ\text{C}$) and the reactor showed deposits on the stirrer, the thermocouple and the wall above the liquid phase.

Example 5

Modification of the supported metallocene catalyst by pre-polymerization

A modified thin-film evaporator was used for the pre-polymerisation. The evaporator was placed in a glove box and equipped with a powder flask that contained the catalyst powder. The system was connected with a second flask filled with oxygen and moisture free 4-methyl-1-pentene. 10 g of the catalyst prepared in example 3 were placed in the powder flask. The pressure was reduced to the vapour pressure of the monomer. 0.1 bar hydrogen was added. The polymerization conversion was followed by weighing the flask. After a polymerization time of one hour the weight ratio pre-polymer to catalyst ($m_{\text{pre-pol}}/m_{\text{cat}}$) was 0.4. Softening of the pre-polymer began at a temperature of 234°C.

Example 6

Propylene polymerization using the supported metallocene catalyst

A 20 l stirred reactor was, after being made inert with

- 26 -

nitrogen, charged at room temperature with 6.5 kg of liquid propylene, 2 l hydrogen and with a 1 molar triethylaluminum/hexane solution and the mixture was stirred for 15 minutes at 200 rpm. The triethylaluminum/hexane solution was fed in such an amount that a TEAL/Zr molar ratio of 600 was adjusted.

260 mg of the unmodified supported metallocene catalyst prepared according to Example 3 were mixed with the co-catalyst for 15 minutes. Subsequently the mixture was rinsed into the reactor using a further 500 g of propylene, the stirrer speed was increased to 400 rpm, the mixture was heated over a period of 12 minutes to the polymerization temperature of 70°C and the temperature was kept constant. The reaction was stopped after two hours by flashing off the propylene. This gave 1536 g of polypropylene with a fines content (<100 μm) determined by sieve analysis, of 5 % by weight. The activity was 5.9 kg / g cat_{pure} x 2h. The polymer particles were granular ($M_w = 345$ kg/mol; polydispersity 2.9; $T_m = 153^\circ\text{C}$) and the reactor showed deposits on stirrer, thermocouple and the wall above the liquid phase.

Example 7

Propylene polymerization using a 4-methyl-1-pentene pre-polymerized catalyst

The polymerisation was carried out by the method described in Example 6, but using 445 mg of the pre-polymerized catalyst of Example 5.

This gave 1938 g of polypropylene with a fines content (<100 μm) determined by sieve analysis of 1.2 % by weight. The activity was 4.4 kg/g cat_{pre-pol} x 2h (6.1 kg/g cat_{pure} x 2h). The polymer particles were granular ($M_w = 324$ kg/mol; polydispersity 2.8; $T_m = 153^\circ\text{C}$) and the

- 27 -

reactor showed no deposits on the internal fittings or the wall.

Example 8

Propylene polymerization using a 4-methyl-1-pentene pre-polymerized catalyst

The pre-polymerization was carried out by the method described in Example 5. The weight ratio pre-polymer to catalyst ($m_{\text{pre-pol}}/m_{\text{cat}}$) was 0.8. Softening of the pre-polymer began at a temperature of 232°C.

The polymerization reactor was charged with 295 mg of the pre-polymerized catalyst.

This gave 1663 g of polypropylene with a fines content (<100 μm) determined by sieve analysis of 1.5 % by weight. The activity was 5.6 kg/g $\text{cat}_{\text{pre-pol}}$ x 2h (10.1 kg/g cat_{pure} x 2h). The polymer particles were granular ($M_w = 355 \text{ kg/mol}$; polydispersity 2.8; $T_m = 152^\circ\text{C}$) and the reactor showed no deposits on the internal fittings or the wall.

Example 9

Propylene polymerization using a 3-methyl-1-butene pre-polymerized catalyst

The pre-polymerisation was carried out by the method described in Example 5. The gaseous 3-methyl-1-butene was directly fed without dilution to the pre-polymerization reactor over a period of 4 hours. No hydrogen was used. The weight ratio pre-polymer to catalyst ($m_{\text{pre-pol}}/m_{\text{cat}}$) was 0.2. Softening of the pre-polymer began at a temperature of 301°C.

The polymerisation reactor was charged with 546 mg of the pre-polymerized catalyst.

- 28 -

This gave 1184 g of polypropylene with a fines content ($<100\ \mu\text{m}$) determined by sieve analysis of 0.9 % by weight. The activity was $2.2\ \text{kg/g cat}_{\text{pre-pol}} \times 2\text{h}$ ($2.6\ \text{kg/g cat}_{\text{pure}} \times 2\text{h}$). The polymer particles were granular ($M_w = 336\ \text{kg/mol}$; polydispersity 2.8; $T_m = 153^\circ\text{C}$) and the reactor showed no deposits on the internal fittings or the wall.

Example 10

Propylene copolymerization using a 4-methyl-1-pentene pre-polymerized catalyst

The pre-polymerization was carried out by the method described in Example 5. The weight ratio pre-polymer to catalyst ($m_{\text{pre-pol}}/m_{\text{cat}}$) was 0.8. Softening of the pre-polymer began at a temperature of 232°C .

The 20 l stirred polymerization reactor was, after being made inert, charged at room temperature with 6.5 kg of liquid propylene, 85 g ethylene, 2 l hydrogen and with a 1 molar triethylaluminum/hexane solution and the mixture was stirred 15 minutes at 200 rpm. The triethylaluminum/hexane solution was fed in such an amount that a TEAL/Zr mole ratio of 600 was adjusted.

The polymerization reactor was charged with 295 mg of the pre-polymerized catalyst.

The reaction was terminated after 30 minutes to limit the extent of conversion. The polymerization run gave 607 g of polypropylene copolymer with a fines content ($<100\ \mu\text{m}$) determined by sieve analysis of 1.6 % by weight. The activity was $5.8\ \text{kg/g cat}_{\text{pre-pol}} \times 2\text{h}$ ($10.4\ \text{kg/g cat}_{\text{pure}} \times 2\text{h}$). The polymer particles were granular ($M_w = 280\ \text{kg/mol}$; polydispersity 2.8; $T_m = 142^\circ\text{C}$) and the reactor showed no deposits on the internal fittings or the wall.

- 29 -

Example 11

Mono-site catalyst based on $\text{Me}_2\text{Si}(2\text{-Me-4-Ph-2-Ind})_2\text{ZrCl}_2$ and MAO

96 g $\text{Me}_2\text{Si}(2\text{-Me-4-Ph-2-Ind})_2\text{ZrCl}_2$ (as can be obtained from Boulder Scientific Company) and 8.6 kg toluene solution of 30 wt% MAO (as can be obtained from Albermarle) were mixed until solution and activation of the metallocene complex, and added slowly to 7 kg of silica, grade Sylopol 2104 (obtained from Grace), calcined at 800°C. The amount of liquid was less than the silica pore volume, hence the silica powder remained free floating and stirrable. The formed mono-site catalyst was dried under nitrogen to a toluene residual amount of 0.45 wt%, having bulk density of 330 kg/m³.

Example 12

Prepolymerization using propylene

333 g of supported catalyst, Example 11, was placed in an inerted glass vessel, volume 3000 mL, equipped with a Teflon bladed stirrer. The vessel was connected to a steel cylinder containing a preweighed amount of liquid propylene. The catalyst vessel nitrogen atmosphere was replaced by gaseous propylene, and pressure from the propylene cylinder adjusted to 0.6 bar. The pressure was increased up to 1.5 bar during 3 hours 15 min at ambient temperature, until the propylene cylinder was emptied. Weight of resulting prepolymerised catalyst was 424 g. The procedure was repeated for another two batches of catalyst, giving overall 1274 g prepolymerised catalyst from 963 g of pure catalyst.

- 30 -

Example 13

Prepolymerisation using 4-Me-1-pentene

332 g of supported catalyst, Example 11, was placed in an inerted glass vessel, volume 3000 mL, equipped with a Teflon bladed stirrer. The vessel was connected to another glass bottle containing 4-Me-1-pentene. The catalyst vessel was evacuated twice, each time refilled with gaseous 4-Me-1-pentene boiling off from the bottle, to substitute the nitrogen atmosphere. The 4-Me-1-pentene pressure in the catalyst vessel was then kept at 0.5 to 0.7 bar at ambient temperature for 4 hours, resulting in 449 g of prepolymerised catalyst. The procedure was repeated for another two batches of catalyst, giving overall 1330 g prepolymerised catalyst from 972 g of pure catalyst.

Example 14

Polymerisation of non batch prepolymerised catalyst in Example 11

The catalyst prepared in Example 11 was dispersed to a 153 g/l oil/grease mixture (80+20 vol% of "Parafinium Liquidum 68" from Technol and "Vaselinfett Album" from Technol). The catalyst mud was then fed to a continuous process of two reactors in series: bulk propylene prepolymerisation (9 min residence time at 20°C) and loop bulk propylene polymerisation (1.9 h residence time at 70°C), 585 ppm of hydrogen and 0.1 g/kgpropylene of triethyl aluminum were present.

The catalyst productivity became 6.2 kg/gCat, or 3.2 kg/gCat*h. Process operability was bad. Inspection of the polymerisation reactor inner surface after 38 hours of operation showed sheeting and plugging from this catalyst. The powder samples, taken out of the lots and from the reactor, contained flat irregular shaped

- 31 -

agglomerates with a diameter up to 1.5 cm. They appeared as sintered particles containing also fines. After reaching the target loop density these agglomerates caused loop reactor discharge problems.

Transparency and partly fused surface of particles were also observed. Product polymer analyses showed MFR = 30.5, XS = 1.25 w%, $T_m = 152.1^\circ\text{C}$. Polymer powder bulk density was 0.456 g/ml.

Example 15

Polymerisation of propylene prepolymerised catalyst in Example 12

The catalyst prepared in Example 12 was dispersed to a 148.5 g/l oil/grease mixture (80+20 vol% of "Parafinium Liquidum 68" from Technol and "Vaselinfett Album" from Technol). The catalyst mud was then fed to a continuous process of two reactors in series: bulk propylene prepolymerisation (9 min residence time at 20°C) and loop bulk propylene polymerisation (1.3 h residence time at 70°C), 620 ppm of hydrogen and 0.1 g/kgpropylene of triethyl aluminum were present.

The catalyst productivity became $4.8 \text{ kg/gCat}_{\text{pure}}$, or $3.5 \text{ kg/gCat}_{\text{pure}} \cdot \text{h}$. Process operability was bad, although less severe than in Example 14. Inspection of the polymerisation reactor inner surface after 46 hours of operation showed plugging from this catalyst, but no sheeting and only few fine particle residues. The powder samples, taken out of the lots and from the reactor, contained flat irregular shaped agglomerates with a diameter up to 1.5 cm. They appeared as sintered particles containing also fines. After reaching the target loop density these agglomerates caused loop reactor discharge problems.

- 32 -

Transparency and partly fused surface of particles were also observed. Product polymer analyses showed MFR = 38, XS = 1.6 wt%. Polymer powder bulk density was 0.434 g/ml.

Example 16

Polymerisation of 4-Me-1-pentene prepolymerised catalyst in Example 13

The catalyst prepared in Example 13 was dispersed to a 150 g/l oil/grease mixture (80+20 vol% of "Parafinium Liquidum 68" from Technol and "Vaselinfett Album" from Technol). The catalyst mud was then fed to a continuous process of two reactors in series: bulk propylene prepolymerisation (9 min residence time at 15°C) and loop bulk propylene polymerisation (1.4 h residence time at 70°C), 670 ppm of hydrogen and 0.1 g/kgpropylene of triethyl aluminum were present.

The catalyst productivity became 6.36 kg/gCat_{pure}, or 4.54 kg/gCat_{pure}*h. Process operability was good, and morphology of powder particles, taken out of the lots and from the reactor, showed neither the transparency nor partly fused surface as noticed while using non batch prepolymerised catalyst in Example 14. Inspection of the polymerisation reactor inner surface after 3.5 days of operation did not show sheeting or plugging from this catalyst. Product polymer analyses showed MFR = 28.3, XS = 1.18 wt%, T_m = 151.7°C. Polymer powder bulk density was 0.416 g/m.

Example 17

Propylene polymerisation using a non batch prepolymerized catalyst

110 mg of the catalyst prepared in Example 11 was placed in a 2l inerted steel reactor, and 0.5 mmol TEAL (as 1 M

- 33 -

heptane solution) and 0.24 bar hydrogen added. Thereafter 1300 ml liquid propylene was added, and 15°C temperature maintained for 8 min. Temperature was then raised to 70°C during 2 min, and maintained for 60 min. Polymerisation was terminated by flashing off the residual propylene. A yield of 340 g polypropylene was obtained. Activity was 4180 g/gCat_{pure}. There was no sheeting in the reactor, but a thin almost invisible sticky layer on reactor walls. Powder morphology was acceptable, although some agglomerates and a little fines. Polymer analyses showed MFR = 110, T_m = 152.0°C, XS = 1.1 wt%.

Example 18

Prepolymerisation using 1-hexene

10.08 g of supported catalyst, Example 11, was placed in an inerted glass vessel, volume 250 ml, equipped with a Teflon bladed stirrer. Pressure was reduced to 100 mbar and 1-hexene in a connected glass bottle was allowed to evaporate into the vessel through a steel tubing. The 1-hexene pressure in the vessel was ca 150 mbar at 25°C. The 1-hexene connection was closed after 143 min, and the residual 1-hexene removed by evacuation. The weight of prepolymerised catalyst was 11.07 g.

Example 19

Propylene polymerisation using a 1-hexene prepolymerized catalyst

200 mg of the catalyst prepared in Example 18 was placed in a 2l inerted steel reactor, and 1.0 mmol TEAL (as 1 M heptane solution) and 0.2 bar hydrogen added. Thereafter 1300 mL liquid propylene was added, and 15°C temperature maintained for 8 min. Temperature was then raised to 70°C during 2 min, and maintained for 60 min.

- 34 -

Polymerisation was terminated by flashing off the residual propylene. A yield of 305 g polypropylene was obtained. Activity was 1670 g/gCat_{pure}. There was no deposits or sheeting in the reactor. Powder morphology was still good, although slightly more agglomerates but very little fines as compared to Example 17.

Polymer analyses showed MFR = 26, $T_m = 152.4^\circ\text{C}$, XS=1.1 wt%.

Example 20

Prepolymerisation using 3-Me-1-butene

1.00 g of supported catalyst, Example 11, was placed in an inerted glass flask, volume 50 ml, equipped with a magnetic stirrer. A steel bottle containing 3-Me-1-butene (obtainable from Aldrich) was connected using a needle through a septum. A pressure reduction valve was placed in between to control the pressure of 3-Me-1-butene. The flask was first flushed with 3-Me-1-butene to substitute the nitrogen, and then pressure was adjusted to 1.3 bar for 1 hour. The weight increase of the glass flask content was 0.25 g. Thereafter the steel bottle was tilted upside down, and 0.55 g of 3-Me-1-butene added as liquid. The 3-Me-1-butene connection was removed, and the glass flask was stored over the week-end without weight loss. Then the residual 3-Me-1-butene was removed by evacuation at 180 torr for 10 min. The weight of prepolymerised catalyst was 1.51 g.

Example 21

Prepolymerisation using 3,3-Me₂-1-butene

1.03 g of supported catalyst, Example 11, was placed in an inerted glass flask, volume 50 ml, equipped with a magnetic stirrer. A glass bottle containing 3,3-Me₂-1-butene (obtainable from Aldrich) was cooled in

- 35 -

a refrigerator to -20°C , purged with nitrogen to remove traces of oxygen, and stored overnight on 13X molecular sieve. While using a syringe through a septum, 3,3-Me₂-1-butene was added batchwise to the catalyst containing glass flask until 0.78 g was accumulated. The glass flask was stored over the week-end without weight loss. Residual 3,3-Me₂-1-butene was removed by evacuation at 185 torr for 10 min. The weight of prepolymerised catalyst was 1.22 g.

Example 22

Propylene polymerisation using a 3-Me-1-butene prepolymerized catalyst

160 mg of the catalyst prepared in Example 20 was placed in a 2l inerted steel reactor, and 0,5 mmol TEAL (as 1 M heptane solution) and 0.24 bar hydrogen added.

Thereafter 1300 ml liquid propylene was added, and 15°C temperature maintained for 8 min. Temperature was then raised to 70°C during 2 min, and maintained for 60 min. Polymerisation was terminated by flashing off the residual propylene. A yield of 410 g polypropylene was obtained. Activity was 3870 g/gCat_{pure}. There were no deposits or sheeting in the reactor. Powder morphology was good, less agglomerates and only a little fines as compared to Example 17.

Polymer analyses showed MFR = 55, $T_m = 152.0^{\circ}\text{C}$, XS=1.2 wt%.

Example 23

Propylene polymerisation using a 3,3-Me₂-1-butene prepolymerized catalyst

130 mg of the catalyst prepared in Example 21 was placed in a 2l inerted steel reactor, and 0,5 mmol TEAL (as 1 M heptane solution) and 0.25 bar hydrogen added.

Thereafter 1300 ml liquid propylene was added, and 15°C

- 36 -

temperature maintained for 8 min. Temperature was then raised to 70°C during 2 min, and maintained for 60 min. Polymerisation was terminated by flashing off the residual propylene. A yield of 355 g polypropylene was obtained. Activity was 3230 g/gCat_{pure}. There were no deposits or sheeting in the reactor. Powder morphology was good, still some agglomerates but very little fines as compared to Example 17.

Polymer analyses showed MFR = 80, T_m = 152.0°C, XS=1.1 wt%.

Example 24

Prepolymerisation using 4-Me-1-pentene on a Ziegler-Natta MgCl₂-supported catalyst

The Ziegler-Natta MgCl₂-supported catalyst was synthesised following procedures given in patent EP-A-591224, essentially by spray crystallizing a magnesium chloride alcoholate followed by treatment with hot titanium tetrachloride, reacting with an 1,2-dicarboxylic ester, washing in a hydrocarbon and drying.

The catalyst was prepolymerised in an oil mud inside an inerted 50 ml glass bottle containing a magnetic stirrer. An amount of 0.300 g catalyst was dispersed into 14.66 oil, and 0.9 ml 0.5 M heptane solution of TEAL was added. Then 0.31 g of neat 4-Me-1-pentene was added, and the mud was stored over night. This oil mud was used for polymerisation without further treatment. Total weight of mud was 16.00 g, giving a cat_{pure} concentration of 1.875 wt%.

Example 25

Propylene polymerisation using a 4-Me-1-pentene prepolymerized Z-N catalyst

1.02 g of the catalyst mud prepared in Example 24 was

- 37 -

placed in a 2l inerted steel reactor, and 1.75 mmol TEAL (as 0.5 M heptane solution), 0.18 mmol cyclohexylmethyldimethoxysilane (as 0.2 M heptane solution), and 0.31 bar hydrogen was added. Thereafter 1300 ml liquid propylene was added, and 15°C temperature maintained for 8 min. Temperature was then raised to 70°C during 2 min, and maintained for 60 min.

Polymerisation was terminated by flashing off the residual propylene. A yield of 300 g polypropylene was obtained. Activity was 15690 g/gCat_{pure}. There were no deposits or sheeting in the reactor. Powder morphology was good, less fines than in Example 26.

Polymer analyses showed MFR = 3.5, T_m = 162.7°C, XS=1.6 wt%.

Example 26

Propylene polymerisation using a non batch prepolymerised Z-N catalyst

The catalyst used for prepolymerisation in Example 24 was dispersed into a 2.00 w% oil mud. 0.89 g of this oil mud was placed in a 2l inerted steel reactor, and 3.6 mmol TEAL (as 0.5 M heptane solution), 0.18 mmol cyclohexylmethyldimethoxysilane (as 0.2 M heptane solution), and 0.30 bar hydrogen was added. Thereafter 1300 mL liquid propylene was added, and 15°C temperature maintained for 8 min. Temperature was then raised to 70°C during 2 min, and maintained for 60 min.

Polymerisation was terminated by flashing off the residual propylene. A yield of 260 g polypropylene was obtained. Activity was 14610 g/gCat_{pure}. There were no deposits or sheeting in the reactor. Powder morphology was acceptable, some fines.

Polymer analyses showed MFR = 5.8, XS=1.7 w%.

Example 27

Prepolymerisation using 4-Me-1-pentene on a Ziegler-Natta silica-supported catalyst

The Ziegler-Natta silica-supported catalyst was synthesised following procedures given in patent EP-A-949274, essentially by precipitating magnesium chloride onto silica, reacting with titanium tetrachloride and evaporating the slurry medium.

The catalyst was prepolymerised in a oil mud inside an inerted 50 ml glass bottle containing a magnetic stirrer. 1.206 g catalyst and 1.25 mmol TEA (as 0.5 M heptane solution) were dispersed in oil to a total weight of 12.00 g. Then 1.0 g of neat 4-Me-1-pentene was added, and the mud was stored over night. This oil mud was used for polymerisation without further treatment. Total weight of mud was 13.0 g, giving a catpure concentration of 9.28 w%.

Example 28

Ethylene polymerisation using a 4-Me-1-pentene prepolymerized Z-N catalyst

1.10 g of the catalyst mud prepared in Example 27 was placed in a 2l inerted steel reactor, and 1.0 mmol TEAL (as 1 M heptane solution) added. A hydrogen pressure of 1.00 bar was added on top of 1.47 bar nitrogen pressure. Thereafter 650 mL liquid isobutane was added, and temperature stabilised at 80°C. Ethylene was then continuously fed, and temperature and pressure controls adjusted to 29.7 bar total pressure at 90°C.

Polymerisation was terminated after 47 min by closing the ethylene feed and flashing off the hydrocarbons. A yield of 425 g polyethylene was obtained. Activity was 4163 g/gCat. There were some deposits and sheeting in the reactor. Powder morphology was good, no fines.

Polymer analyses showed $MFR_2 = 0.39$, $MFR_{21} = 12.6$,

- 39 -

density = 0.9576 g/cm³.

Example 29

Ethylene polymerisation using a non batch prepolymerised Z-N catalyst

98 mg of the catalyst used for prepolymerisation in Example 27 was mixed with 1.0 mmol TEAL (as 1 M heptane solution) and placed in a 2l inerted steel reactor, and 1.02 bar hydrogen was added on top of 1.47 bar nitrogen pressure. Thereafter 650 ml liquid isobutane was added, and temperature stabilised at 80°C. Ethylene was then continuously fed, and temperature and pressure controls adjusted to 30.7 bar total pressure at 90°C.

Polymerisation was terminated after 60 min by closing the ethylene feed and flashing off the hydrocarbons. A yield of 100 g polyethylene was obtained. Activity was 1020 g/gCat. There were no significant deposits or sheeting in the reactor. Powder morphology was good, although some agglomerates, no fines.

Polymer analyses showed $MFR_2 = 0.18$, $MFR_{21} = 5.4$, density = 0.9549 g/cm³.

Claims:

1. A modified supported olefin polymerization catalyst, characterised in that the catalyst is modified by pre-polymerization with an olefin or an olefin mixture different from the olefin or olefin mixture of the subsequent olefin polymerization.
2. The modified supported catalyst as claimed in claim 1 wherein the melting point or softening point of the polyolefin or olefin copolymer resulting from pre-polymerization is at least 20°C, preferably 30°C higher than the melting point or softening point of the polyolefin or olefin copolymer of the subsequent olefin polymerization.
3. The modified supported catalyst as claimed in claim 1 being a metallocene or Ziegler-Natta catalyst.
4. The modified supported catalyst as claimed in claim 3 comprising two or more metallocenes forming a multi-site catalyst.
5. The modified supported catalyst as claimed in claim 1 wherein the pre-polymerization olefin or olefin mixture is one or more of propylene, 1-butene, 3-methyl-1-butene, 3,3-dimethyl-1-butene, 4-methyl-1-pentene, 4,4-dimethyl-1-pentene, 4,5-dimethyl-1-hexene, vinyl cyclopentane, vinyl cyclohexane, vinyl cycloheptane, allyl cyclopentane, allyl cyclohexane or allyl cycloheptane.
6. The modified supported catalyst as claimed in claim 5 wherein the pre-polymerization olefin or olefin mixture is one or more of 3-methyl-1-butene, 4-methyl-1-

- 41 -

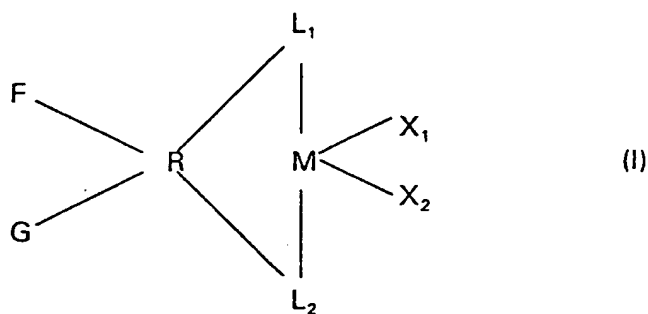
pentene, vinyl cyclopentane or vinyl cyclohexane.

7. The modified supported catalyst as claimed in claim 1 wherein the support material is silica and/or magnesium chloride.

8. The modified supported catalyst as claimed in claim 3 being a supported metallocene catalyst, comprising

A) 90.0 - 99.9 parts by weight of a catalyst support based on a hydrophilic inorganic oxide of an element of main groups II to IV or transition group IV of the Periodic Table or a mixture or mixed oxide thereof, which catalyst support is obtainable by simultaneous reaction with aluminoxanes and with polyfunctional organic crosslinkers,

B) 10 - 0.1 parts by weight of a metallocene compound of the formula I



where

M is a metal selected from the group consisting of Ti, Zr, Hf, V, Nb and Ta or an element selected from the group consisting of the lanthanides,

X₁ and X₂ are identical or different and are each a C₁-C₁₀-alkyl group, a C₁-C₁₀-alkoxy group, a C₆-C₁₀-aryl group, a C₆-C₁₀-aryloxy group, a C₂-C₁₀-alkenyl group, a C₇-C₂₀-alkylaryl group, a C₇-C₂₀-arylalkyl group, a C₈-C₂₀-arylalkenyl group, hydrogen or a halogen atom,

L₁ and L₂ are identical or different and are each an

- 42 -

unsubstituted, monosubstituted or polysubstituted monocyclic or polycyclic hydrocarbon radical containing at least one cyclopentadienyl unit which can form a sandwich structure with M,

R is carbon, silicon, germanium or tin,

F and G are identical or different and are each a trimethylsilyl radical of the formula $-\text{Si}(\text{CH}_3)_3$, where G may also be a C_1 - C_{10} -alkyl radical, preferably a C_1 - C_4 -alkyl radical, or a C_6 - C_{10} -aryl radical,

characterized in that the supported metallocene catalyst comprising A and B, has an activity of maximum 100 kg, preferably maximum 50 kg, particularly preferred maximum 20 kg polyolefin/g supported catalyst/h.

9. A process for preparing the modified supported catalyst as claimed in any preceding claim, comprising the step of modifying a supported olefin polymerization catalyst by pre-polymerization with an olefin or an olefin mixture different from the olefin or olefin mixture of the subsequent olefin polymerization.

10. The process for preparing a modified supported catalyst as claimed in claim 9, characterized in that the pre-polymerization is carried out in gas phase.

11. The process for preparing a modified supported catalyst as claimed in claim 9, characterized in that the pre-polymerization is slurry phase polymerization in an oil or grease slurry medium.

12. The process for preparing a modified supported catalyst as claimed in claim 9, characterized in that for liquid-phase polymerization of propylene the modification of the supported catalyst is carried out by pre-polymerization of 4-methyl-1-pentene.

13. A process for preparing polyolefins by polymerization or copolymerization of olefins, wherein a modified supported catalyst as claimed in any of claims 1 to 8 is contacted with the olefins.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/GB 00/03775

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C08F10/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C08F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 94 28034 A (HOECHST AG ; EXXON CHEMICAL PATENTS INC (US)) 8 December 1994 (1994-12-08) cited in the application page 10, line 17 -page 15, line 18; claim 2 ---	1-5,7,9, 11,13
X Y	US 5 618 886 A (SHINOZAKI TETSUNORI ET AL) 8 April 1997 (1997-04-08) column 20, line 15 -column 20, line 60; claim 1; examples 1-41 ---	1-3,5-7, 9,11-13 10
X	EP 0 321 218 A (MITSUI PETROCHEMICAL IND) 21 June 1989 (1989-06-21) page 8, line 8 -page 8, line 24; claims 1-11; examples 1,2 --- -/--	1-3,5-7, 9,11,13



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *G* document member of the same patent family

Date of the actual completion of the international search

29 November 2000

Date of mailing of the international search report

06/12/2000

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Van Golde, L

INTERNATIONAL SEARCH REPORT

International Application No

PCT/GB 00/03775

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 517 183 A (HIMONT INC ;MONTECATINI TECNOLOGIE SRL (IT)) 9 December 1992 (1992-12-09) claim 1; examples 1-3 ----	1-3, 5, 7, 9, 11, 13
Y	US 5 688 734 A (TORMASCHY KELLY ET AL) 18 November 1997 (1997-11-18) the whole document ----	10
A	EP 0 787 746 A (DANUBIA PETROCHEM POLYMERE) 6 August 1997 (1997-08-06) cited in the application the whole document -----	8-13

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/GB 00/03775

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9428034 A	08-12-1994	AU 683899 B	27-11-1997
		AU 7041394 A	20-12-1994
		CN 1130912 A,B	11-09-1996
		DE 69408989 D	16-04-1998
		DE 69408989 T	05-11-1998
		EP 0705281 A	10-04-1996
		ES 2117279 T	01-08-1998
		FI 955687 A	24-01-1996
		JP 8511044 T	19-11-1996
		NO 954780 A	15-01-1996
		RU 2126017 C	10-02-1999
		ZA 9403595 A	14-12-1995
US 5618886 A	08-04-1997	CA 2130009 A,C	14-02-1995
		CN 1111643 A,B	15-11-1995
		CN 1232829 A	27-10-1999
		CN 1231293 A	13-10-1999
		CN 1231295 A	13-10-1999
		CN 1231294 A	13-10-1999
		CN 1239105 A	22-12-1999
		CN 1245813 A	01-03-2000
		EP 0641807 A	08-03-1995
		EP 0854156 A	22-07-1998
		JP 7102028 A	18-04-1995
		KR 182339 B	15-05-1999
		KR 163667 B	15-01-1999
		KR 163666 B	15-01-1999
		KR 163665 B	15-01-1999
		JP 7102029 A	18-04-1995
		JP 7109309 A	25-04-1995
		JP 7109314 A	25-04-1995
		JP 7109313 A	25-04-1995
EP 0321218 A	21-06-1989	JP 1156305 A	19-06-1989
		JP 2529310 B	28-08-1996
		JP 1156353 A	19-06-1989
		JP 8026190 B	13-03-1996
		AT 110746 T	15-09-1994
		CA 1306579 A	18-08-1992
		CN 1036776 A,B	01-11-1989
		CN 1063444 A,B	12-08-1992
		DE 3851313 D	06-10-1994
		DE 3851313 T	22-12-1994
		ES 2063053 T	01-01-1995
		KR 9201040 B	01-02-1992
		US 5159021 A	27-10-1992
		US 5266636 A	30-11-1993
EP 0517183 A	09-12-1992	IT 1246614 B	24-11-1994
		AT 141094 T	15-08-1996
		AU 653829 B	13-10-1994
		AU 1917792 A	08-01-1993
		BR 9205273 A	20-07-1993
		CA 2088681 A	04-12-1992
		CN 1069034 A	17-02-1993
		CZ 9203960 A	16-06-1993
		DE 69212633 D	12-09-1996
		DE 69212633 T	20-02-1997

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/GB 00/03775

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0517183 A		DK 541760 T	02-09-1996
		WO 9221706 A	10-12-1992
		EP 0541760 A	19-05-1993
		ES 2090646 T	16-10-1996
		FI 930151 A	14-01-1993
		GR 3020753 T	30-11-1996
		HU 66659 A,B	28-12-1994
		IL 102074 A	24-01-1995
		JP 6503123 T	07-04-1994
		KR 218863 B	01-09-1999
		MX 9202633 A	01-12-1992
		NO 180721 B	24-02-1997
		PL 297841 A	13-12-1993
		RU 2114125 C	27-06-1998
		ZA 9203985 A	24-02-1993
US 5688734 A	18-11-1997	CA 2223258 A	23-01-1997
		CN 1192752 A	09-09-1998
		EP 0836624 A	22-04-1998
		JP 11508932 T	03-08-1999
		WO 9702297 A	23-01-1997
EP 0787746 A	06-08-1997	AT 403919 B	25-06-1998
		DE 19608804 A	18-09-1997
		AT 18896 A	15-11-1997
		AT 183207 T	15-08-1999
		BR 9700826 A	25-08-1998
		CA 2196509 A	03-08-1997
		CZ 9700195 A	13-08-1997
		DE 59700318 D	16-09-1999
		DK 787746 T	06-12-1999
		ES 2135958 T	01-11-1999
		HU 9700315 A	02-03-1998
		JP 9216909 A	19-08-1997
		NO 970421 A	04-08-1997
		SK 13497 A	07-10-1998
		US 5932514 A	03-08-1999

**This Page is Inserted by IFW Indexing and Scanning
Operations and is not part of the Official Record**

BEST AVAILABLE IMAGES

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images include but are not limited to the items checked:

- ☒ **BLACK BORDERS**
- ☐ **IMAGE CUT OFF AT TOP, BOTTOM OR SIDES**
- ☐ **FADED TEXT OR DRAWING**
- ☐ **BLURRED OR ILLEGIBLE TEXT OR DRAWING**
- ☐ **SKEWED/SLANTED IMAGES**
- ☐ **COLOR OR BLACK AND WHITE PHOTOGRAPHS**
- ☐ **GRAY SCALE DOCUMENTS**
- ☐ **LINES OR MARKS ON ORIGINAL DOCUMENT**
- ☐ **REFERENCE(S) OR EXHIBIT(S) SUBMITTED ARE POOR QUALITY**
- ☐ **OTHER:** _____

IMAGES ARE BEST AVAILABLE COPY.

As rescanning these documents will not correct the image problems checked, please do not report these problems to the IFW Image Problem Mailbox.